



The Biotechnology Education Company®



**273**  
EDVO-Kit #

## Radial Immunodiffusion

**Storage:**

Store the entire experiment  
in the refrigerator.

**EXPERIMENT OBJECTIVES:**

Radial Immunodiffusion is a sensitive quantitative technique that is often used clinically to detect patient levels of blood proteins. In this experiment, students will learn to quantitatively determine the unknown concentration of an antigen.

All components are intended for educational research only. They are not to be used for diagnostic or drug purposes, nor administered to or consumed by humans or animals.

The Biotechnology Education Company® • 1-800-EDVOTEK • [www.edvotek.com](http://www.edvotek.com)

## Table of Contents

---

	Page
Experiment Components	3
Experiment Requirements	3
Background Information	
Radial Immunodiffusion	4
Experiment Procedures	
Experiment Overview	6
Student Experimental Procedures	7
Study Questions	11
Instructor's Guidelines	
Notes to the Instructor	13
Pre-Lab Preparations	14
Avoiding Common Pitfalls	16
Expected Results	17
Study Questions and Answers	18
Material Safety Data Sheets	19



## Experiment Components

- A Antibody Solution
  - B Standard Antigen Solution
  - C UltraSpec-Agarose™
  - D Buffer Powder
  - E Unknown Concentration of Antigen
- 
- 1 Sleeve Petri Dishes
  - 2 10 ml Pipets
  - 10 Well Cutters
  - 80 Transfer Pipets
  - 70 Microtest Tubes
  - 1 Graph Paper Template
  - 1 Practice Loading Solution

This experiment is  
designed for  
10 groups.

Store entire  
experiment in  
the refrigerator.

## Requirements

- Automatic Micropipets and Tips (5-50  $\mu$ l)
- Pipet Pumps (for 10 ml pipets)
- Ruler
- Plastic Box or Dish
- Plastic Wrap
- Foil
- Paper Towels
- Distilled Water
- Heat plate, Bunsen burner, or microwave
- 400 to 600 ml beaker or Erlenmeyer flask
- 150 ml beaker or flask
- Water bath
- 250 ml Graduated Cylinder
- 37°C Incubation Oven

All components are intended for educational research only. They are not to be used for diagnostic or drug purposes, nor administered to or consumed by humans or animals.

## Radial Immunodiffusion

The fundamental reaction of immunology involves the interaction of antibodies (Ab) and antigens (Ag). These interactions are useful in the defense of the body against bacterial and viral infections and toxins. The defense capabilities are dependent upon the recognition of antigens by humoral components of the immune system. Specific antibodies are then produced in response to exposure to the antigen.

The formation of antigen-antibody complexes is the first step in removing infectious agents from the body. Because each antibody can bind more than one antigen and each antigen can be bound by more than one antibody molecule, very large macromolecular complexes can form. These complexes form precipitates which can be cleared from the body through various means. These precipitates are also useful for laboratory and diagnostic tests.

When antibodies and antigens are inserted into different areas of an agarose gel, they diffuse toward each other and form opaque bands of precipitate at the interface of their diffusion fronts. Precipitation reactions of antibodies and antigens in agarose gels provide a method of analyzing the various antibody-antigen reactions in a system.

Double diffusion in two dimensions is a simple procedure invented by the Swedish scientist, Ouchterlony. Antigen and antibody solutions are placed in separate wells cut in an agarose plate. The reactants diffuse from the wells toward each other and precipitate where they meet at equivalent proportions. A single antigen will combine with its homologous antibody to form a single precipitation line.

Radial immunodiffusion (RID) is a technique that can quantitatively determine the concentration of an antigen. Unlike many gel and liquid precipitation techniques which qualitatively detect antigen, RID is a sensitive quantitative technique that is often used clinically to detect patient levels of blood proteins.

Antibody is incorporated into molten agarose which is poured into a Petri dish and allowed to solidify. Small wells are cut into the agarose and are filled with known concentrations of antigen which corresponds to the antibody in the agarose. Samples of unknown concentrations are placed in similar wells. The antigens in solution then diffuse outwards from the well in a circular pattern surrounding the well. Antibody is present in excess and diffusion of the antigen will continue until a stable ring of antigen-antibody precipitate forms. There are antigen-antibody complexes

EDVOTEK and The Biotechnology Education Company are registered trademarks of EDVOTEK, Inc.



Duplication of this document, in conjunction with use of accompanying reagents, is permitted for classroom/laboratory use only. This document, or any part, may not be reproduced or distributed for any other purpose without the written consent of EDVOTEK, Inc. Copyright © 1992, 1993, 1994, 1996, 1997, 1998, 2005 EDVOTEK, Inc., all rights reserved. EVT 004185AM

## Radial Immunodiffusion

throughout the zone surrounding the well within the precipitin line. At the precipitin line is where the greatest number of complexes can be found because the antigen and antibody are present in roughly equal proportions. This is known as the equivalence zone or equivalence point. Generally, it takes 24 to 48 hours for optimal diffusion to occur and precipitation to become apparent.

For each antigen, an endpoint precipitation ring of a certain diameter will form. From the known standard concentrations, a standard curve can be drawn by plotting antigen concentration versus the diameter squared measurements of the rings. From this linear calibration curve the concentration of the unknown antigen samples may be determined.

## Experiment Overview

### EXPERIMENT OBJECTIVE:

Radial Immunodiffusion is a sensitive quantitative technique that is often used clinically to detect patient levels of blood proteins. In this experiment, students will learn to quantitatively determine the unknown concentration of an antigen.



Wear gloves  
and safety  
goggles

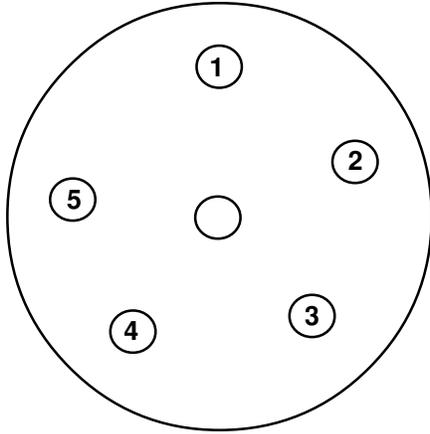
### LABORATORY SAFETY

1. Gloves and goggles should be worn routinely as good laboratory practice.
2. DO NOT MOUTH PIPET REAGENTS - USE PIPET PUMPS OR BULBS.
3. Dispose of RID plates through proper laboratory waste disposal procedures.



Duplication of this document, in conjunction with use of accompanying reagents, is permitted for classroom/ laboratory use only. This document, or any part, may not be reproduced or distributed for any other purpose without the written consent of EDVOTEK, Inc. Copyright © 1992, 1993, 1994, 1996, 1997, 1998, 2005 EDVOTEK, Inc., all rights reserved. EVT 004185AM

## Student Experimental Procedures



Circle shown actual size.

Dilution	Concentration
Undiluted	2 mg/ml
1:2	1 mg/ml
1:4	0.5 mg/ml
1:8	0.25 mg/ml
1:16	0.125 mg/ml

### A. PREPARATION OF AGAROSE PLATES

1. Place the template under the plate so the pattern is centered.
2. Cut the wells using the well cutter (provided in the kit) in a gentle punching motion. Remove the agarose plugs with a flat-edged toothpick or spatula.

### B. PREPARING THE STANDARDS (SERIAL DILUTION)

1. Label four microtest tubes: 1:2, 1:4, 1:8, and 1:16.
2. Using a micropipet, add 50 microliters of Buffer to each tube.
3. With a fresh pipet tip, add 50 microliters of "Standard" to the tube labeled 1:2. Mix.
4. With a fresh pipet tip, transfer 50 microliters of the 1:2 dilution to the tube labeled 1:4. Mix.
5. With a fresh pipet tip, transfer 50 microliters of the 1:4 dilution to the tube labeled 1:8. Mix.
6. With a fresh pipet tip, transfer 50 microliters of the 1:8 dilution to the tube labeled 1:16. Mix.
7. There are now five antigen samples for the standard curve (see chart).

## Student Experimental Procedures

### C. PRACTICE WELL LOADING (OPTIONAL)

This experiment contains practice loading solution. This solution is included to allow instructors and students to practice loading the sample wells before performing the actual experiment. Use a micropipetting device or one of the plastic transfer pipets included in the experiment to practice loading the sample wells with the practice loading solution. Make enough copies of the template for each lab group.

1. One practice plate should be prepared for every two groups. Enough reagents have been provided for this purpose.
2. Using the well cutters provided, cut several wells in the agarose as shown in the template below. Refer to Student Instructions for preparation of sample wells.
3. Practice loading the sample wells with the Practice Loading Solution using a micropipetting device. Load 5  $\mu\text{l}$  per well and make sure the sample covers the entire surface of the well. If a micropipetting device is not available, use the transfer pipets provided, taking care not to overfill the wells. If using transfer pipets, put in just enough sample to cover the bottom of the well.

### D. LOADING THE SAMPLES

1. On the bottom of the plate, number the wells on the perimeter of the plate 1 through 5. Leave the center well unlabeled.
2. Load wells 1 through 5 using the same pipet tip or transfer pipet. In well #5, load 5  $\mu\text{l}$  of the 1:16 antigen dilution. Make sure the sample covers the entire surface of the well by carefully spreading it with the pipet tip.
3. In well #4, load 5  $\mu\text{l}$  of the 1:8 antigen dilution.
4. In well #3, load 5  $\mu\text{l}$  of the 1:4 antigen dilution.
5. In well #2, load 5  $\mu\text{l}$  of the 1:2 antigen dilution.
6. In well #1, load 5  $\mu\text{l}$  of the undiluted antigen.
7. With a fresh pipet tip or microtipped transfer pipet, load 5  $\mu\text{l}$  of your unknown in the center well.
8. Label the cover of the Petri dish with your lab group number or your initials. Place the cover on the dish, place the dish right side up (do not invert) inside the incubation chamber on the paper toweling. Cover the incubation chamber and place in a 37°C incubation oven or at room temperature for 24 to 48 hours.

#### Remember!



You may use the same pipet tip or transfer pipet to load wells #1 through #5, starting with the most dilute antigen dilution and ending with the most concentrated. Use a fresh tip for the unknown.



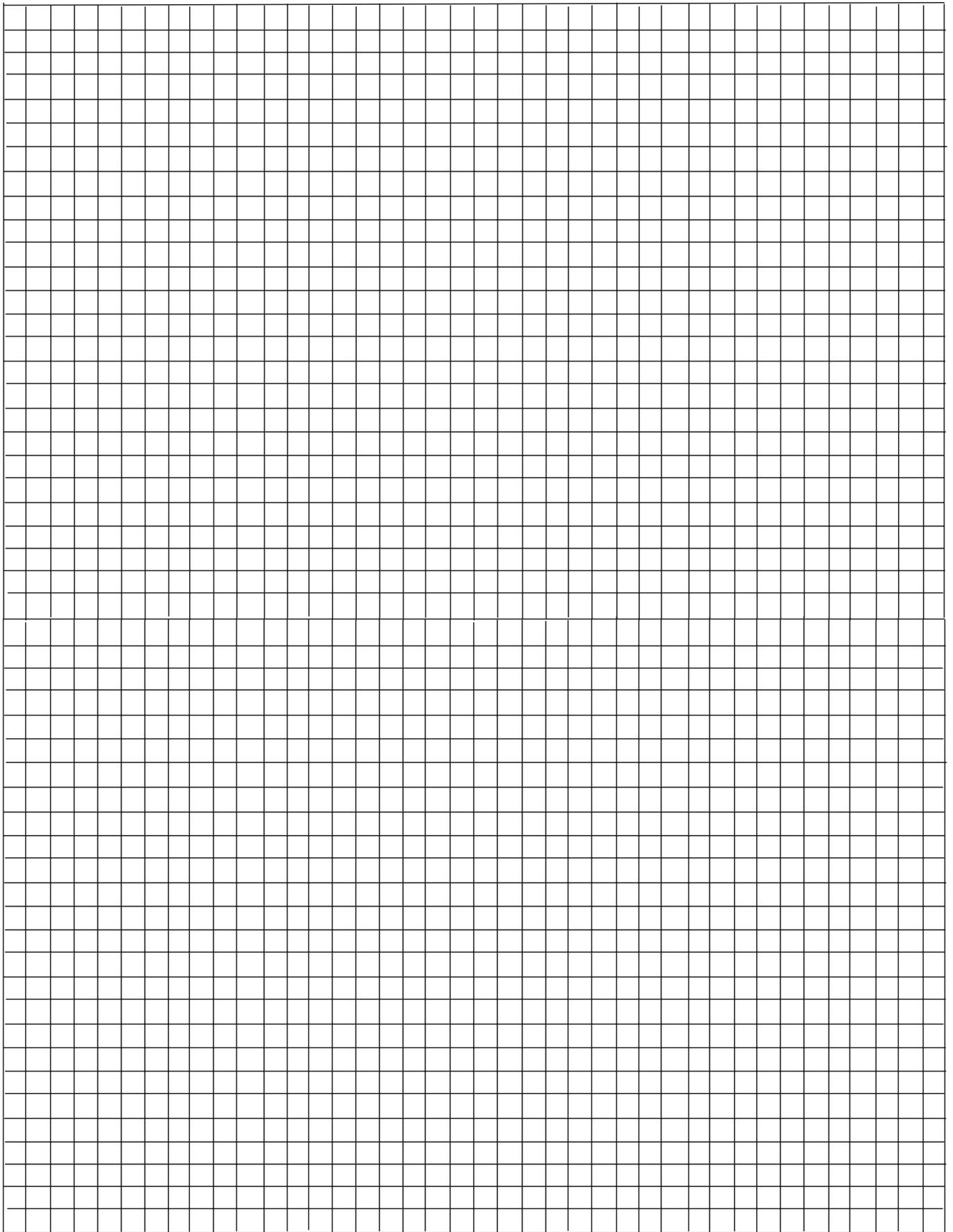
Duplication of this document, in conjunction with use of accompanying reagents, is permitted for classroom/ laboratory use only. This document, or any part, may not be reproduced or distributed for any other purpose without the written consent of EDVOTEK, Inc. Copyright © 1992, 1993, 1994, 1996, 1997, 1998, 2005 EDVOTEK, Inc., all rights reserved. EVT 004185AM

## Student Experimental Procedures

### E. READING THE RESULTS

The precipitin rings will be visible in 24 to 48 hours. Carefully hold a plate up so that the overhead room lights shine through it. You should be able to see opaque circles around each well where antigen and antibody have precipitated.

With a ruler, measure the diameter (through the centers of the wells) of the precipitin ring in millimeters. To plot the standard curve, square the diameter value and plot antigen concentration on the X-axis and the diameter squared on the Y-axis. Draw the best fit line through these points. Calculate the value of the unknown antigen concentration from this graph.



## Experiment Results and Study Questions

### LABORATORY NOTEBOOK RECORDINGS:

Address and record the following in your laboratory notebook or on a separate worksheet.

#### Before starting the experiment:

- Write a hypothesis that reflects the experiment.
- Predict experimental outcomes.

#### During the Experiment:

- Record (draw) your observations, or photograph the results.

#### Following the Experiment:

- Formulate an explanation from the results.
- Determine what could be changed in the experiment if the experiment were repeated.
- Write a hypothesis that would reflect this change.

### STUDY QUESTIONS

Answer the following study questions in your laboratory notebook or on a separate worksheet.

1. What do the circular precipitin rings represent?
2. Why do the ring sizes change until equilibrium is reached?
3. Predict the results if a very low concentration of antigen were loaded into a well. What would happen if not enough antibody was incorporated into the agarose?
4. Compare and contrast Radial Immunodiffusion with its close relative, the Ouchterlony plate technique.

Notes:

---



Duplication of this document, in conjunction with use of accompanying reagents, is permitted for classroom/ laboratory use only. This document, or any part, may not be reproduced or distributed for any other purpose without the written consent of EDVOTEK, Inc. Copyright © 1992, 1993, 1994, 1996, 1997, 1998, 2005 EDVOTEK, Inc., all rights reserved. EVT 004185AM

## Notes to the Instructor

If you do not find the answers to your questions in this section, a variety of resources are continuously being added to the EDVOTEK® web site. In addition, Technical Service is available from 9:00 am to 6:00 pm, Eastern time zone. Call for help from our knowledgeable technical staff at 1-800-EDVOTEK (1-800-338-6835).

### APPROXIMATE TIME REQUIREMENTS

Your individual schedule and time requirements will determine when the RID plates should be prepared. It takes approximately 30 minutes to prepare the plates (generally 10 minutes of this time is required for solidification). Students can prepare the plates, if time allows.

#### Online Ordering



Visit our web site for information about EDVOTEK®'s complete line of experiments for biotechnology and

#### Technical Service Department

Mon - Fri  
9:00 am to 6:00 pm ET



FAX: (301) 340-0582  
web: [www.edvotek.com](http://www.edvotek.com)  
email: [edvotek@aol.com](mailto:edvotek@aol.com)

Please have the following information:

- The experiment number and title
- Kit Lot number on box or tube
- The literature version number (in lower right corner)
- Approximate purchase date

Duplication of this document, in conjunction with use of accompanying reagents, is permitted for classroom/laboratory use only. This document, or any part, may not be reproduced or distributed for any other purpose without the written consent of EDVOTEK, Inc. Copyright © 1992, 1993, 1994, 1996, 1997, 1998, 2005 EDVOTEK, Inc., all rights reserved.

EVT 004185AM



## PreLab Preparations



Goggles and gloves should be worn during all steps involving heat.

**PREPARE AGAROSE IN BUFFER**

1. In a 400 to 600 ml beaker or Erlenmeyer flask, add entire contents of buffer powder package (component D) to 200 ml distilled water. Swirl the flask until the powder is in solution. Remove 50 ml for use as dilution buffer to a separate beaker.
2. Add the entire contents of agarose package (component C) to the flask or beaker containing 150 ml of buffer. Swirl to disperse large clumps. With a marking pen, indicate the level of solution volume on the outside of the flask or beaker.
3. The solution must be boiled to dissolve the agarose. This can be accomplished with a hot plate or microwave. Cover the beaker with foil and heat the mixture to boiling over the burner with occasional swirling. Wear safety goggles and use hot gloves. Boil until all the agarose is dissolved. Check to make sure that there are no small, clear particles of agarose. The final solution should be clear.

Heat the mixture to dissolve the agarose powder. The final solution should be clear (like water) without any undissolved particles.

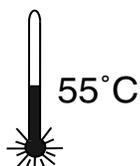
## A. Microwave method:

- Cover flask with plastic wrap to minimize evaporation.
- Heat the mixture on High for 1 minute.
- Swirl the mixture and heat on High in bursts of 25 seconds until all the agarose is completely dissolved.

## B. Hot plate method:

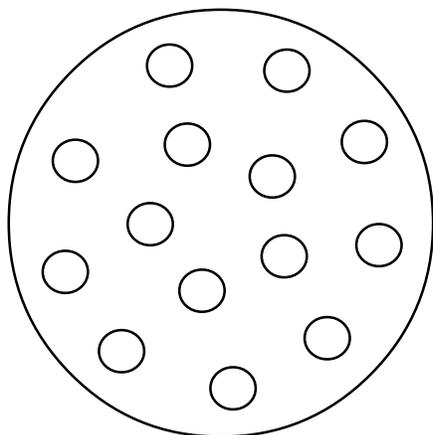
- Cover the flask with foil to prevent excess evaporation.
- Heat the mixture to boiling over a burner with occasional swirling. Boil until all the agarose is completely dissolved.

4. If detectable evaporation has occurred, add hot distilled water to adjust the volume of solution up to the original level as marked on the flask or beaker in step 2. Do not use cool water, or the agarose solution may cool too quickly and prematurely solidify.
5. Cool the agarose solution to 55°C in a waterbath. Swirl occasionally while cooling.



Duplication of this document, in conjunction with use of accompanying reagents, is permitted for classroom/ laboratory use only. This document, or any part, may not be reproduced or distributed for any other purpose without the written consent of EDVOTEK, Inc. Copyright © 1992, 1993, 1994, 1996, 1997, 1998, 2005 EDVOTEK, Inc., all rights reserved. EVT 004185AM

## Pre-Lab Preparations



Practice Plate  
Circle shown actual size.

### EACH STUDENT GROUP SHOULD RECEIVE:

- 1 tube Buffer
- 1 tube Standard
- 1 tube Unknown
- 4 microtest tubes
- 1 practice plate
- 1 experimental RID plate
- 1 well cutter
- 1 template
  
- Micropipetting device and tips (or 8 transfer pipets)
- Graph paper
- Ruler
- Marking pen

### PREPARATION OF PRACTICE PLATES

1. If practice plates are to be made, pipet 2.5 ml of molten agarose into each of 10 petri dishes with a 10 ml pipet. Gently spread the agarose with the pipet on the bottom of the plate to cover the entire surface. Return the remaining agarose to waterbath.
2. Allow agarose plates to set up and cool. Refrigerate if plates are not to be used within a few hours.

### PREPARATION OF ANTIBODY PLATES

1. Pour 26 ml of molten agarose solution to a large tube or flask.
2. Add entire contents of Antibody Solution (Component A) to the 26 ml warm agarose solution. With a pipet, stir the solution to mix. Keep the solution warm (such as in the 55°C waterbath) so it does not prematurely set up. The antibody concentration will be 1 mg/ml.
3. With a 10 ml pipet, dispense 2.5 ml into the bottom of each Petri dish, gently spread the agarose with the pipet to cover the bottom. Allow the agarose to solidify. This will take approximately 10 minutes. If the plates are not to be used the day of the preparation, they can be wrapped in plastic wrap and stored in the refrigerator for no longer than one week.
4. Each group requires 1 antibody plate and 1 practice plate.

## Pre-Lab Preparations

### PREPARATION OF ANTIGENS

Students will prepare serial dilutions of the Standard Antigen Solution (Component B) to determine the standard curve.

1. Label 10 microtest tubes with "Standard".
2. Label 10 microtest tubes with "Unknown".
3. Label 10 microtest tubes with "Buffer".
4. Aliquot 75  $\mu$ l of Standard Antigen Solution (Component B) into each tube labeled "Standard".
5. Aliquot 10  $\mu$ l of Unknown Antigen Solution (Component E) into each tube labeled "Unknown".
6. Aliquot 1 ml of Buffer (retained from plate preparation step) into each tube labeled "Buffer".
7. Each group requires one tube each of Standard, Unknown, and Buffer.

### PREPARATION OF INCUBATION CHAMBER

#### (Prepare Day of Laboratory)

1. Obtain plastic container or dish with lid. If a lid is not available, the container may be covered with plastic wrap.
2. Line the bottom of the container with several paper towels. Add distilled water to the towels to saturate. There should not be any liquid above the paper towels. All the liquid should be absorbed into the towels. Cover the chamber with the lid or plastic wrap.

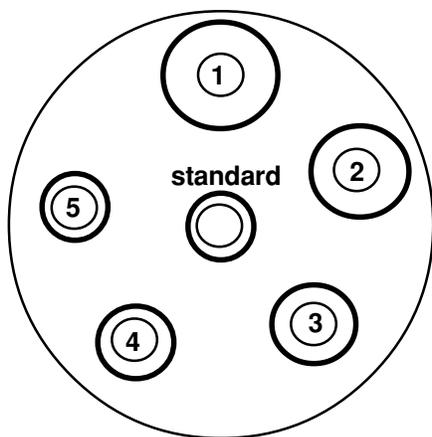


Duplication of this document, in conjunction with use of accompanying reagents, is permitted for classroom/laboratory use only. This document, or any part, may not be reproduced or distributed for any other purpose without the written consent of EDVOTEK, Inc. Copyright © 1992, 1993, 1994, 1996, 1997, 1998, 2005 EDVOTEK, Inc., all rights reserved. EVT 004185AM

## Avoiding Common Pitfalls

1. Follow instructions carefully when preparing gels. Make sure the agarose is completely dissolved.
2. Make neat, clean wells with the well cutters. Take measures to ensure that the wells are properly spaced according to the template on page 5.
3. Add samples to the wells carefully and precisely. Avoid overfilling the wells.
4. Do not tip or invert plates when transferring to the humidity chamber.
5. Placing the humidity chamber in a 37° C incubation oven will expedite the formation of precipitin arcs.

## Expected Results



On regular graph paper, a linear standard curve should be obtained. If the curve is not linear, the unknown concentration cannot be accurately determined. Precipitin rings will vary based on the concentration of the antigens, antibody, agarose and the time and temperature of incubation.

From the standard curve, the unknown concentration can be determined by finding the diameter squared value on the Y-axis, finding the intersecting point on the standard curve line, and obtaining the value on the X-axis. The value on the X-axis is the concentration of antigen in the solution. The concentration provided was 0.40 mg/ml.

**Please refer to the kit  
insert for the Answers to  
Study Questions**

 <b>Material Safety Data Sheet</b> May be used to comply with OSHA's Hazard Communication Standard. 29 CFR 1910.1200 Standard must be consulted for specific requirements.			
IDENTITY (As Used on Label and List) Agarose		Note: Blank spaces are not permitted. If any item is not applicable, or no information is available, the space must be marked to indicate that.	
<b>Section I</b>			
Manufacturer's Name <b>EDVOTEK, Inc.</b> Address (Number, Street, City, State, Zip Code) <b>14676 Rothgeb Drive            Rockville, MD 20850</b>		Emergency Telephone Number <b>(301) 251-5990</b> Telephone Number for information <b>(301) 251-5990</b> Date Prepared 09-15-2002 Signature of Preparer (optional)	
<b>Section II - Hazardous Ingredients/Identify Information</b>			
Hazardous Components [Specific Chemical Identity; Common Name(s)] OSHA PEL ACGIH TLV Other Limits Recommended % (Optional)			
This product contains no hazardous materials as defined by the OSHA Hazard Communication Standard. CAS #9012-36-6			
<b>Section III - Physical/Chemical Characteristics</b>			
Boiling Point For 1% solution	194° F	Specific Gravity (H <sub>2</sub> O = 1)	No data
Vapor Pressure (mm Hg.)	No data	Melting Point	No data
Vapor Density (AIR = 1)	No data	Evaporation Rate (Butyl Acetate = 1)	No data
Solubility in Water Insoluble - cold			
Appearance and Odor White powder, no odor			
<b>Section IV - Physical/Chemical Characteristics</b> N.D. = No data			
Flash Point (Method Used)	No data	Flammable Limits	LEL N.D. UEL N.D.
Extinguishing Media Water spray, dry chemical, carbon dioxide, halon or standard foam			
Special Fire Fighting Procedures Possible fire hazard when exposed to heat or flame			
Unusual Fire and Explosion Hazards None			

<b>Section V - Reactivity Data</b>			
Stability	Unstable		Conditions to Avoid
	Stable	X	None
Incompatibility No data available			
Hazardous Decomposition or Byproducts			
Hazardous Polymerization	May Occur		Conditions to Avoid
	Will Not Occur	X	None
<b>Section VI - Health Hazard Data</b>			
Route(s) of Entry: Inhalation? Yes Skin? Yes Ingestion? Yes			
Health Hazards (Acute and Chronic) Inhalation: No data available Ingestion: Large amounts may cause diarrhea			
Carcinogenicity: NTP? IARC Monographs? OSHA Regulation?			
Signs and Symptoms of Exposure No data available			
Medical Conditions Generally Aggravated by Exposure No data available			
Emergency First Aid Procedures Treat symptomatically and supportively			
<b>Section VII - Precautions for Safe Handling and Use</b>			
Steps to be Taken in case Material is Released for Spilled Sweep up and place in suitable container for disposal			
Waste Disposal Method Normal solid waste disposal			
Precautions to be Taken in Handling and Storing None			
Other Precautions None			
<b>Section VIII - Control Measures</b>			
Respiratory Protection (Specify Type) Chemical cartridge respirator with full facepiece.			
Ventilation	Local Exhaust		Special
	Mechanical (General)/Gen. dilution ventilation		Other
Protective Gloves	Yes	Eye Protection	Splash proof goggles
Other Protective Clothing or Equipment Impervious clothing to prevent skin contact			
Work/Hygienic Practices None			

 <b>Material Safety Data Sheet</b> May be used to comply with OSHA's Hazard Communication Standard. 29 CFR 1910.1200 Standard must be consulted for specific requirements.			
IDENTITY (As Used on Label and List) 10x PBS		Note: Blank spaces are not permitted. If any item is not applicable, or no information is available, the space must be marked to indicate that.	
<b>Section I</b>			
Manufacturer's Name <b>EDVOTEK, Inc.</b> Address (Number, Street, City, State, Zip Code) <b>14676 Rothgeb Drive            Rockville, MD 20850</b>		Emergency Telephone Number <b>(301) 251-5990</b> Telephone Number for information <b>(301) 251-5990</b> Date Prepared 09-19-2002 Signature of Preparer (optional)	
<b>Section II - Hazardous Ingredients/Identify Information</b>			
Hazardous Components [Specific Chemical Identity; Common Name(s)] OSHA PEL ACGIH TLV Other Limits Recommended % (Optional)			
N/A Blend			
<b>Section III - Physical/Chemical Characteristics</b>			
Boiling Point	100°C	Specific Gravity (H <sub>2</sub> O = 1)	1.017
Vapor Pressure (mm Hg.)	No data	Melting Point	No data
Vapor Density (AIR = 1)	No data	Evaporation Rate (Butyl Acetate = 1)	No data
Solubility in Water soluble			
Appearance and Odor colorless liquid			
<b>Section IV - Physical/Chemical Characteristics</b>			
Flash Point (Method Used)	Noncombustible	Flammable Limits	LEL UEL
Extinguishing Media Use extinguishing media appropriate to surrounding fire			
Special Fire Fighting Procedures Wear SCBA and protective clothing to prevent contact with skin and eyes			
Unusual Fire and Explosion Hazards Emits toxic fumes under fire conditions			

<b>Section V - Reactivity Data</b>			
Stability	Unstable		Conditions to Avoid
	Stable		
Incompatibility Strong acids			
Hazardous Decomposition or Byproducts Nature of decomposition products not known			
Hazardous Polymerization	May Occur		Conditions to Avoid
	Will Not Occur		
<b>Section VI - Health Hazard Data</b>			
Route(s) of Entry: Inhalation? Yes Skin? Yes Ingestion? Yes			
Health Hazards (Acute and Chronic) Cause eye & skin irritation, material is irritating to mucous membranes and upper respiratory tract. The toxicological properties have not been thoroughly investigated.			
Carcinogenicity: NTP? IARC Monographs? OSHA Regulation?			
Signs and Symptoms of Exposure			
Medical Conditions Generally Aggravated by Exposure			
Emergency First Aid Procedures Swallowed - wash out mouth with water provided person is conscious. Skin/eye contact - flush with water Inhalation - remove to fresh air			
<b>Section VII - Precautions for Safe Handling and Use</b>			
Steps to be Taken in case Material is Released for Spilled Wear respirator, chemical safety goggles, rubber boots and heavy rubber gloves, sweep up, place in a bag and hold for waste disposal.			
Waste Disposal Method For small quantities - cautiously add to a large stirred excess of water. Adjust pH to neutral			
Precautions to be Taken in Handling and Storing Wear appropriate NIOSH/MSHA approved respirator, chemical resistant gloves, safety goggles safety shower and eye bath.			
Other Precautions			
<b>Section VIII - Control Measures</b>			
Respiratory Protection (Specify Type) NIOSH/MSHA approved respirator			
Ventilation	Local Exhaust	N/A	Special N/A
	Mechanical (General)	N/A	Other N/A
Protective Gloves	Yes	Eye Protection	Yes
Other Protective Clothing or Equipment			
Work/Hygienic Practices Do not ingest. Avoid contact with skin, eyes and clothing. Wash thoroughly after handling.			